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Soil Micronutrients-Unsaid

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Abstract

Micronutrients are essential plant mineral nutrients taken up and utilised by fruit crops in very small quantities. They constitute in total less than 1% of dry weight of most plants. Micronutrients increase the quality and yield of fruit trees. Besides they play multiple roles in fruit development for example formation of growth hormones, chlorophyll formation, oxidation-reduction reactions, colour development, etc. Plant growth and development may be retarded if any of these elements is lacking within soil or is not adequately balanced with other nutrients. Factors such as soil pH, organic matter, temperature, soil type and moisture are important determining factors for the availability of micronutrients (Fageria *et al.*, 2001). In recent years however deficiencies of micronutrients have been diagnosed more frequently by farmers due to greater amount of intensive cropping, high demand of modern crop cultivars, losses through leaching and so on (Scherer, 2001). Micronutrient deficiency is usually corrected by addition of micronutrient containing compounds such as, sodium tetraborate (14-20% B), boric acid (17%B), ferrous ammonium sulphate (14% Fe), zinc sulphate (23-36% Zn), copper sulphate (13-35% Cu), potassium chloride (47% Cl), manganese sulphate (23-25% Mn), ammonium molybdate (54% Mo), nickel chloride (25% Ni) (Lohry, 2007). Without adequate micronutrients, crops can't reach their full potential in terms of both yield as well as quality. Highest yield of pistachio was produced in two levels of Zn and maximum vegetative growth took place in Zn and Fe treatment, whereas highest of splitting rate was detected in combination of Cu and Fe and lowest one in control, but effect of Zn, Cu and Fe on blankness was not significant (Soliemanzadeh *et al.*, 2013). Since micronutrients are involved in various enzymatic activities, their deficiency causes malfunctioning of plant activities. Response to micronutrients varies from crop to crop and depending upon the soil type and its status. In addition to yield it improves quality of fruit (Bahmanyar and Mashae, 2010).

Keywords: Micronutrients, chlorophyll formation, oxidation-reduction reactions, colour development, etc.

Introduction

One hundred and nine elements have been identified and included in the periodic table so far. Most of these exist in the periodic table so far. Most of these exist in earth's mantle, earth's crust and soil, though the magnitude of their occurrence varies. To a certain extent plant composition reflects the chemical composition of the soil. A nutrient element is one that is required to complete the life cycle of the organism and its relative deficiency produces specific deficiency symptoms. The adverse effects are relieved by the supply of that specific element only. Elements essential for the plants may not be essential for the other organisms and vice versa. Depending upon availability, the nutrient content in a plant could be deficient, sufficient or toxic. Nutrient content is considered deficient when it severely limits growth and produces characteristic deficiency symptoms. Nutrient contents associated with only growth reductions and not accompanied by appearance of deficiency symptoms are termed insufficient. Range of nutrient content in plants associated with optimum crop yields is called sufficient. When the concentration of a nutrient element rises too high to cause significant growth reductions, it is termed toxic. (Rattan and Goswami, 2012)^[28].

Essential elements

Elements needed by the plant without which it will not be able to survive are called as essential nutrients. For an element to be regarded as an essential nutrient, it must satisfy the following criteria, as propounded by Arnon and Stout (1939):

1. A deficiency of an essential nutrient element makes it impossible for the plant to complete the vegetative or reproductive stage of its life-cycle.
2. The deficiency is specific to the element and can be prevented or corrected only by supplying that element.

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- The element is involved directly in the nutrition of the plant, quite apart from its possible effects in correcting some microbiological or chemical conditions of the soil or other culture medium.

The seventeen nutrients recognised essential for the plant growth are: carbon(C), hydrogen (H), oxygen (O), nitrogen (N), phosphorus (P), potassium (K), calcium (Ca), magnesium (Mg), sulphur (S), iron (Fe), manganese (Mn), zinc (Zn), copper (Cu), boron (B), molybdenum (Mo), chlorine (Cl), nickel (Ni). Nickel is the last addition to the list of essential elements, done in 1987.

Depending upon the quantity required by the plants nutrients are classified into macronutrients and micronutrients.

Macronutrients

Macronutrients or major nutrients are so called because these are required in large quantities, more than that of iron. These include C, H, O, N, P, K, Ca, Mg, and S. Carbon, hydrogen and oxygen constitute 90 to 95 % of the plant drymatter weight and are supplied through carbon dioxide (CO₂) and water. Remaining six major nutrients, viz. N, P, K, Ca, Mg and S are further subdivided into primary and secondary nutrients. Primary nutrients: Nitrogen, phosphorus and potassium are termed as primary nutrients because of their larger requirement by the plants and correction of their wide spread deficiencies is often necessary through application of commercial fertilizers of which these are the major constituents.

Secondary nutrients

Calcium, magnesium and sulphur are termed as secondary nutrients because of their moderate requirements by plants, localised deficiencies and their inadvertent alleviation by incidental accretions. For example, the phosphatic fertilizers, single superphosphate (SSP) contains Ca and S. Likewise ammonium sulphate, a nitrogenous fertilizer also supplements S.

Micronutrients

Micronutrients are elements which are essential for plant growth, but are required in much smaller amounts than those of the primary nutrients; nitrogen, phosphorus and potassium. The micronutrients are boron (B), copper (Cu), iron (Fe), manganese (Mn), molybdenum (Mo), zinc (Zn), and chloride (Cl). While chloride is micronutrient, deficiencies rarely occur in nature, so discussions on supplying micronutrient fertilizers are confined to the other six micronutrients. Deficiencies of micronutrients have been increasing in some crops. Some reasons are higher crop yields which increase plant nutrient demands, use of high analyses NPK fertilizers containing lower quantities of micronutrient contaminants, and decreased use of farmyard manure on many agricultural soils. Micronutrient deficiencies have been verified in many soils through increased use of soil testing and plant analysis.

Micronutrient Nutrition

A brief discussion of micronutrient functions and nutrient deficiency symptoms in plants and soil conditions affecting micronutrient availability serves to help understand their importance in crop production and to recognize symptoms of possible deficiencies:

Boron

A primary function of boron is related to cell wall formation, so boron-deficient plants may be stunted. Sugar transport in

plants, flower retention and pollen formation and germination also are affected by boron. Seed and grain production are reduced with low boron supply. Boron-deficiency symptoms first appear at the growing points. This results in a stunted appearance (rosetting), barren ears due to poor pollination, hollow stems and fruit (hollow heart) and brittle, discoloured leaves and loss of fruiting bodies. Boron deficiencies are mainly found in acid, sandy soils in regions of high rainfall, and those with low soil organic matter. Borate ions are mobile in soil and can be leached from the root zone. Boron deficiencies are more pronounced during drought periods when root activity is restricted.

Copper

Copper is necessary for carbohydrate and nitrogen metabolism, so inadequate copper results in stunting of plants. Copper also is required for lignin synthesis which is needed for cell wall strength and prevention of wilting. Deficiency symptoms of copper are dieback of stems and twigs, yellowing of leaves, stunted growth and pale green leaves that wither easily. Copper deficiencies are mainly reported on organic soils (peats and mucks), and on sandy soils which are low in organic matter. Copper uptake decreases as soil pH increases. Increased phosphorus and iron availability in soils decreases copper uptake by plants.

Iron

Iron is involved in the production of chlorophyll, and iron chlorosis is easily recognized on iron sensitive crops growing on calcareous soils. Iron also is a component of many enzymes associated with energy transfer, nitrogen reduction and fixation, and lignin formation. Iron is associated with sulphur in plants to form compounds that catalyze other reactions. Iron deficiencies are mainly manifested by yellow leaves due to low levels of chlorophyll. Leaf yellowing first appears on the younger upper leaves in interveinal tissues. Severe iron deficiencies cause leaves to turn completely yellow or almost white, and then brown as leaves die. Iron deficiencies are found mainly on calcareous (high pH) soils, although some acid, sandy soils low in organic matter also may be iron-deficient. Cool, wet weather enhances iron deficiencies, especially on soils with marginal levels of available iron. Poorly aerated or compacted soils also reduce iron uptake by plants. Uptake of iron decreases with increased soil pH, and is adversely affected by high levels of available phosphorus, manganese and zinc in soils.

Manganese

Manganese is necessary in photosynthesis, nitrogen metabolism and to form other compounds required for plant metabolism. Interveinal chlorosis is a characteristic manganese-deficiency symptom. In very severe manganese deficiencies, brown necrotic spots appear on leaves, resulting in premature leaf drop. Delayed maturity is another deficiency symptom in some species. Whitish-gray spots on leaves of some cereal crops and shortened internodes in cotton are other manganese-deficiency symptoms. Manganese deficiencies mainly occur on organic soils, high-pH soils, sandy soils low in organic matter, and on over-limed soils. Soil manganese may be less available in dry, well-aerated soils, but can become more available under wet soil conditions when manganese is reduced to the plant-available form. Conversely, manganese toxicity can result in some acidic, high-manganese soils. Uptake of manganese decreases with increased soil pH and is adversely affected by high levels of available iron in soils.

Molybdenum

Molybdenum is involved in enzyme systems relating to nitrogen fixation by bacteria growing symbiotically with legumes. Nitrogen metabolism, protein synthesis and sulphur metabolism are also affected by molybdenum. Molybdenum has a significant effect on pollen formation, so fruit and grain formation are affected in molybdenum-deficient plants. Because molybdenum requirements are so low, most plant species do not exhibit molybdenum-deficiency symptoms. These deficiency symptoms in legumes are mainly exhibited as nitrogen-deficiency symptoms because of the primary role of molybdenum in nitrogen fixation. Unlike the other micronutrients, molybdenum-deficiency symptoms are not confined mainly to the youngest leaves because molybdenum is mobile in plants. The characteristic molybdenum-deficiency symptom in some vegetable crops is irregular leaf blade formation known as whiptail, but interveinal mottling and marginal chlorosis of older leaves also have been observed. Molybdenum deficiencies are found mainly on acid, sandy soils in humid regions. Molybdenum uptake by plants increases with increased soil pH, which is opposite that of the other micronutrients. Molybdenum deficiencies in legumes may be corrected by liming acid soils rather than by molybdenum applications. However, seed treatment with molybdenum sources may be more economical than liming in some areas.

Zinc

Zinc is an essential component of various enzyme systems for energy production, protein synthesis, and growth regulation. Zinc-deficient plants also exhibit delayed maturity. Zinc is not mobile in plants so zinc deficiency symptoms occur mainly in new growth. Poor mobility in plants suggests the need for a constant supply of available zinc for optimum growth. The most visible zinc-deficiency symptoms are short internodes (rosetting) and a decrease in leaf size. Chlorotic bands along the midribs of corn, mottled leaves of dry bean and chlorosis

of rice are characteristic zinc-deficiency symptoms. Loss of lower bolls of cotton and narrow, yellow leaves in the new growth of citrus also have been diagnosed as zinc deficiencies. Delayed maturity also is a symptom of zinc-deficient plants. Zinc deficiencies are mainly found on sandy soils low in organic matter and on organic soils. Zinc deficiencies occur more often during cold, wet spring weather and are related to reduced root growth and activity as well as lower microbial activity decreases zinc release from soil organic matter. Zinc uptake by plants decreases with increased soil pH. Uptake of zinc also is adversely affected by high levels of available phosphorus and iron in soils.

Chloride

Because chloride is a mobile anion in plants, most of its functions relate to salt effects (stomatal opening) and electrical charge balance in physiological functions in plants. Chloride also indirectly affects plant growth by stomatal regulation of water loss. Wilting and restricted, highly branched root systems are the main chloride-deficiency symptoms, which are found mainly in cereal crops. Most soils contain sufficient levels of chloride for adequate plant nutrition. However, reported chloride deficiencies have been reported on sandy soils in high rainfall areas or those derived from low-chloride parent materials.

Nickel

As nickel is taken by plants in nickel ion forms it is associated with nitrogen metabolism by way of influencing urease activity. In systems where urea is used as the sole N fertilizer for foliar spray and Ni-supply is poor, lower urease activity causes urea toxicity to foliage and leads to severe necrosis of the root tips. In free living *Rhizobia*, adequate Ni-supply ensures optimum hydrogenase activity. It facilitates transport of nutrients to the seeds or grains. There is reduction in dry matter, decrease in amino acid content and accumulation of nitrates due to nickel deficiency.

Concentration of micronutrients in plants (dry weight basis)

Table 1

| Element | Symbol | Form | Range of concentration | Adequate concentration |
|------------|--------|-------------------------------------|------------------------|------------------------|
| Boron | B | H ₃ BO ₃ | 0.2-800ppm | 20ppm |
| Chlorine | Cl | Cl ⁻ | 10-80,000ppm | 100ppm |
| Copper | Cu | Cu ⁺ , Cu ²⁺ | 2-50ppm | 6ppm |
| Molybdenum | Mo | MoO ₄ ²⁻ | 0.10-10ppm | 0.1ppm |
| Manganese | Mn | Mn ²⁺ | 10-600ppm | 50ppm |
| Iron | Fe | Fe ³⁺ , Fe ²⁺ | 20-600ppm | 100ppm |
| Zinc | Zn | Zn ²⁺ | 10-250ppm | 20ppm |

Source: TNAU Agri Portal

Micronutrient sources commonly used for correcting deficiencies in plants

Table 2

| Micronutrient | Common fertilizer sources |
|---------------|--|
| B | Sodium tetraborate (14%-20%B), Solubor (20%B), Liquid boron (10%B), Boric acid (17%B) |
| Fe | Ferrous ammonium sulfate (14%Fe), Ferrous ammonium phosphate (29%Fe) |
| Zn | Zinc sulfate(23-36%Zn), Zinc ammonia complex(10%Zn), Zinc oxide (50-80%Zn) Zinc chelate (9-14%Zn) |
| Cu | Copper sulfate(13-35%Cu), Copper oxide(75-89%Cu) |
| Cl | Potassium chloride (47%Cl), Sodium chloride (60%Cl), Ammonium chloride (66%Cl), Calcium chloride (64%Cl), Magnesium chloride (74%Cl) |
| Mn | Manganese sulfate (23-25%Mn), Manganese oxide (41-68%Mn) |
| Mo | Ammonium molybdate (54%Mo), Sodium molybdate (39%Mo), Molybdenum trioxide (66%Mo), Molybdic acid (53%Mo) |
| Ni | Nickel chloride (25%Ni), Nickel nitrate(20%Ni), Nickel oxide(79%Ni) |

Lohry, 2007 [22]

Methods of correcting micronutrient deficiencies

| Element | Corrective measures | |
|---------|--|--|
| | Soil application | Foliar application |
| B | 0.5-2kg borax ha ⁻¹ | 0.15% B |
| Cl | 20-50 kg KCl ha ⁻¹ | Unknown |
| Cu | 5-10 kg CuSO ₄ ha ⁻¹ | 0.1-0.5% CuSO ₄ .5H ₂ O |
| Fe | 30-100 kg FeSO ₄ ha ⁻¹ | 2% FeSO ₄ .7H ₂ O. |
| Mn | 5-50 kg MnSO ₄ ha ⁻¹ | 0.4-0.6% MnSO ₄ |
| Mo | 0.01-1 kg Ammonium Molybdate ha ⁻¹ | 0.07-0.1% Na |
| Zn | 0.5-35 kg ZnSO ₄ | 0.1-0.5% ZnSO ₄ .7H ₂ O |

Importance of micronutrients

1. Increase quality and yield: Most micronutrients act as co-factors in various enzymes taking part in various metabolisms of plants like protein metabolism, carbohydrate metabolism, photosynthesis etc. therefore there will be increase in protein content, total soluble salts (TSS) and other quality parameters, which result in improving the quality. Micronutrients such as iron which is essential for chlorophyll formation and as a result photosynthesis will increase and thus there will be increase in yield as well.
2. In legumes it influences nitrogen fixation: N-fixation takes place using enzyme nitrogenase. This enzyme consists of two proteins- Fe & Mo protein. Reaction occurs while N₂ is fixed in nitrogenase complex. Fe protein first is reduced by electrons donated by ferredoxin. Then this reduced Fe protein binds ATP and reduces Mo-Fe protein, which donates electrons to N₂ producing HN=NH. Further HN=NH is reduced to H₂N=H₂N and this is in turn reduced to 2NH₃.
3. Effect of micronutrient concentrations in planting seed on the vigour of the next season's crop.
4. Major economic impact of micronutrients in a farming operation is through the increased efficiency of macronutrient fertilizer use.

Determining the need for micronutrients

1. Ensure that poor crop growth is not due to macronutrient deficiency or some environmental problems like drought, salinity, disease or insect problems.
2. Look for the micronutrient deficiency in the field.
3. Take soil & plant tissue samples from both affected & unaffected areas and send them for testing.
4. Go for the specific micronutrient application to the crop as per the analysis of the testing.

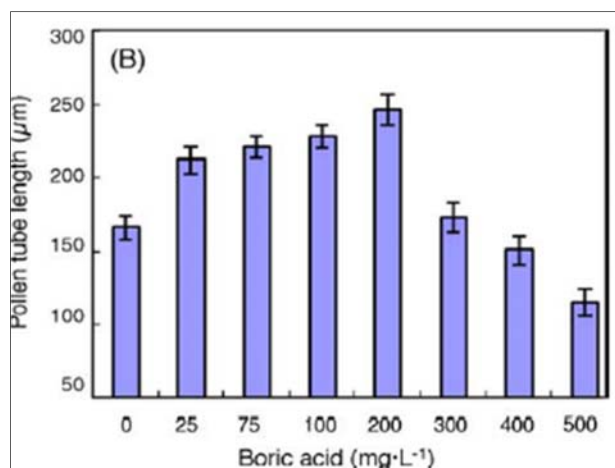
Causes of micronutrient deficiencies

1. Intensive cropping: Crops are grown intensively on a piece of land which results in depletion of micronutrients.
2. High demand of modern crop cultivars: Since there is need to develop new crop cultivars which have high yield potential and high quality parameters to meet the market demand. These modern crop cultivars are exhaustive in nature that is they deplete soil of micronutrients.
3. Loss of top soil by erosion: Due to heavy precipitation, heavy wind which is rich in micronutrient and thus deficiency occurs.
4. Losses of micronutrients through leaching: Excessive rainfall results in leaching of micronutrients to deeper layers of soil. Thus there is deficiency of micronutrient in rhizosphere.

5. Use of marginal lands for crop production: since these lands have less fertility for crop production so micronutrients availability is less.

Factors affecting the availability of micronutrients

1. Soil pH: Most micronutrients are available under acidic conditions except Mo and B. At low pH micronutrient cations have high solubility as a result it becomes available whereas at high pH these micronutrient forms hydroxy ions which finally are changed to insoluble hydroxides and oxides of elements as a result it becomes unavailable. Mo and B get fixed by silicate clays and oxides of Fe and Al as a result become unavailable.
2. Organic matter: It contains organic compounds, these organic compounds react with micronutrients and form chelates which have high solubility and thus availability increases.
3. Temperature: Availability of most micronutrients tends to decrease at low temperature because root activity and microbial activity gets reduced and low rates of diffusion and dissolution of micronutrients increases temperature.
4. Moisture: Low moisture content decreases the micronutrient availability. When moisture content decreases colloidal particles may become immobilised as a result micronutrient adsorption on surface takes place and decreases their availability.



Lee *et al.* 2009 [20]

Fig 1: Effect of boron on pollen tube length in pear.

Results

Boron has been demonstrated to perform a crucial function in the synthesis and/or structure of plant cell walls, via its binding with rhamnogalacturonan II chains, resulting in the formation of the B-polysaccharide complex (Kaneko *et al.*, 1997; Kobayashi *et al.*, 1996) [16, 18]. Boron-complexing capability promotes the migration of B and polysaccharides across biological membranes (Matoh, 1997) [24] to pollen tube wall synthesis sites. Pollen tube precursors are known to be rich in polypeptides, most notably glycoprotein and arabinofuranosyl residue-rich polysaccharides (Li and Liskens, 1983) [21]. Many of these compounds have been determined to form strong complexes with boron (Loomis and Durst, 1992) [23]. Therefore, the inhibitory effects on pollen tube growth associated with high boron contents may be attributable to excessive cross-linking of pollen cell walls (Hu and Brown, 1994). This would imply that moderate levels of boron might stimulate pollen germination and subsequent pollen tube growth in pear pollen.

Table 2: Effect of zinc, copper and iron on yield, nut weight, splitting rate, blank nut rate and vegetative growth of Pistachio.

| Treatments | Yield(dryweight,cluster ⁻¹) | Nut weight(100 nut g ⁻¹) | Splitting rate (%) | Blankness (%) | Vegetative growth (cm) |
|---|---|---------------------------------------|--------------------|---------------|------------------------|
| Control | 8.09 | 90.57bc | 83.03c | 11.53a | 11.55ab |
| Zn ₀ Cu ₀ Fe ₁ | 10.51bc | 89.69c | 85.89bc | 21.27a | 10.45ab |
| Zn ₀ Cu ₁ Fe ₀ | 8.55dc | 91.11bc | 91.86ab | 10.82a | 12.12ab |
| Zn ₀ Cu ₁ Fe ₁ | 10.79ab | 94.13ab | 95.52a | 20.88a | 11.97ab |
| Zn ₁ Cu ₀ Fe ₀ | 12.77a | 94.55ab | 89.25abc | 15.28a | 10.70ab |
| Zn ₁ Cu ₀ Fe ₁ | 8.032d | 95.86a | 82.22c | 14.90a | 11.42ab |
| Zn ₁ Cu ₁ Fe ₀ | 9.12bcd | 92.86abc | 86.88bc | 18.54a | 10.57ab |
| Zn ₁ Cu ₁ Fe ₁ | 8.37cd | 85.53d | 87.21bc | 16.27a | 10.12ab |
| Zn ₂ Cu ₀ Fe ₀ | 10.86ab | 90.82bc | 88.68bc | 20.58a | 9.82b |
| Zn ₂ Cu ₀ Fe ₁ | 7.12d | 84.10de | 81.50c | 20.94a | 13.25a |
| Zn ₂ Cu ₁ Fe ₀ | 7.27d | 93.26abc | 85.19bc | 17.08a | 10.85ab |
| Zn ₂ Cu ₁ Fe ₁ | 8.37cd | 80.92e | 89.75abc | 14.14a | 12.55ab |

Solimanzadeh et al. 2013 [33]

Zn₀=0 mg l⁻¹, Zn₁=1000 mg l⁻¹, Zn₂=2000 mg l⁻¹; Cu₀=0 mg l⁻¹, Cu₁=200 mg l⁻¹; Fe₀=0 mg l⁻¹; Fe₁=300 mg l⁻¹

Results

Perhaps these increases in yield were due to significant increase in leaf Zn and Fe concentrations, which in turn induce more flowering and minimize fruit let drop in pistachio trees. Zn-deficient shoots are markedly smaller in size and much redder than normal. This results indicating foliar spray of Zn were effective in increasing the nut weight of pistachio

trees. Probably zinc does not play an important role in shell splitting of pistachio. This may result from the limited mobility of applied Zn, which has been attributed, at least in part, to the high binding capacity of leaf tissue for Zn (Zhang and Brown 1999) [37]. Blanking can occur during two different phases of pistachio nut development, nut setting and nut filling. It can be affected by crop load and production practices (Ferguson et al. 2005) [11]. The higher vegetative growth for Zn₂Cu₀Fe₁ treatment could be explained by the competition for assimilates between vegetative and fruit growth.

Table 3: Effect of different micronutrients on fruit parameters of peach.

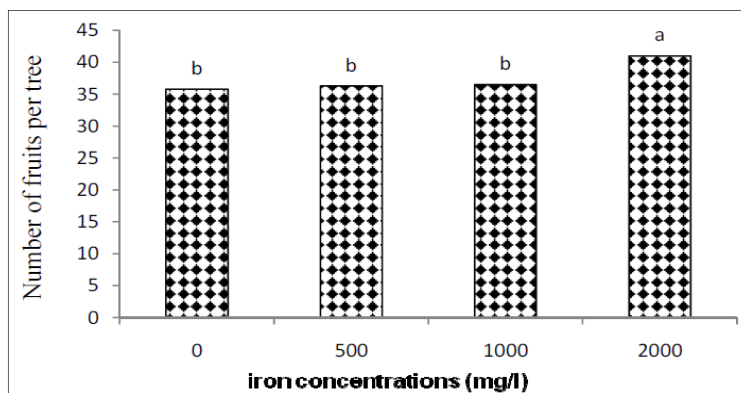
| Treatments | Fruit length (cm) | Fruit diameter (cm) | Fruit volume (ml) | Fruit firmness (lb inch ⁻²) |
|---|-------------------|---------------------|-------------------|---|
| T1(0.1%H3BO3) | 5.38 | 4.61 | 41.68 | 10.97 |
| T2(0.5% ZnSO4.7H2O) | 5.18 | 4.62 | 42.50 | 10.94 |
| T3 (0.5%FeSO4.7H2O) | 5.30 | 4.62 | 41.74 | 10.96 |
| T4(0.1%H3BO3+0.5%ZnSO4.7H2O) | 5.48 | 4.69 | 43.17 | 11.71 |
| T5(0.1%H3BO3+0.5%FeSO4.7H2O) | 5.57 | 4.68 | 43.83 | 11.99 |
| T6(0.5%ZnSO4.7H2O+0.5%FeSO4.7H2O) | 5.25 | 4.68 | 42.53 | 11.29 |
| T7(0.1%H3BO3+0.5%ZnSO4.7H2O+0.5%FeSO4.7H2O) | 5.59 | 5.08 | 44.57 | 12.37 |
| T8(control) | 5.12 | 4.33 | 39.76 | 10.35 |
| CD at 5% | 0.09 | 0.02 | 0.20 | 0.52 |

Yadav et al. 2013 [36]

Results

The increase in size of fruit as a result of foliar application of micronutrients might be because it improved the internal physiology of developing fruit in terms of better supply of water, nutrients, and other compounds vital for their proper growth and development (Dutta and Banik, 2007) [9]. The increase in fruit weight and volume might be due to increase in cell size and intercellular space (Basker and Davis, 1951)

[3]. Zinc has been identified as component of almost 60 enzymes and it has a role in synthesis of growth promoter hormone (auxin). Which is directly associated with improvement of Fresh weight of fruits (Shivanandam et al., 2007) [32]. A favorable effect of foliar application of boron might be due to its role in cell division, cell elongation, sugar metabolism and accumulation of carbohydrates (Sourour, 2000) [34].



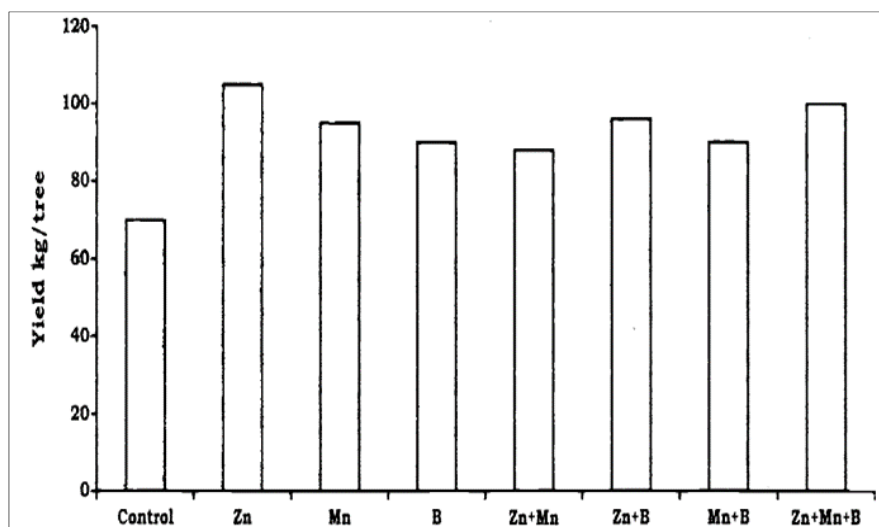
Davarpanah et al. 2013

Fig 2: Effect of different concentrations of iron chelated foliar feeding on the number of fruit per tree in Pomegranate.

Results

Irons have a positive effect on the synthesis and activity of chlorophylls and thereby increase the Photosynthesis. The ability of having more Photosynthesis and food cause to

increase generative power and whereby the tree can hold more fruits. Foliar injecting of Iron sulfate in apple trees increased the amount of fruit in following year (Barney *et al.*, 1984).



Parveen *et al.* 2000

Table 3: Effect of zinc, manganese and boron on the yield of orange.

Results

The main effect Zn shows 13% increase in yield as compared to yield obtained from trees where Zn was not included in foliar spray. This significant increase in yield was due to the significant increase in leaf Zn concentration which in turn induced more flowering and minimize fruitlet drop in sweet oranges. It was reported that fruitlet drop was decreased as leaf Zn and Mn content increased (Garcia *et al.*, 1984) [13]. Foliar application of Mn to trees significantly increased the yield as compared to trees not sprayed with Mn. The optimum yield was obtained from trees sprayed with Mn alone

followed by trees sprayed with Mn+B and these were significantly higher than the control treatment. The increase in yield was due to more fruit setting and less fruit drop in sweet orange. Foliar application of B to trees increased yield significantly as compared to trees not sprayed with B. These results confirm the findings of Chiu and Chang (1986) who reported 3-15% increase in yield due to B sprays. The cumulative effect of Zn and B was found in reducing heavy fruit drop and increased fruit set (Sato, 1962) [29]. It was concluded that foliar spray of Zn, Mn and B corrected the deficiency symptoms and increased the yield.

Table 6: Effect of micronutrient sprays on morphological and reproductive parameters of Aonla.

| Treatment | Plant height (m) | Fruit drop (%) | Fruit retention (%) |
|---|------------------|----------------|---------------------|
| T0 (control) | 0.69 | 79.19 | 20.80 |
| T1 (0.2% Borax) | 0.77 | 74.70 | 25.30 |
| T2 (0.4% Borax) | 0.79 | 64.19 | 35.81 |
| T3 (0.4% Zinc Sulphate) | 0.75 | 78.80 | 21.20 |
| T4 (0.8% Zinc Sulphate) | 0.78 | 69.40 | 30.60 |
| T5 (0.6% Calcium Nitrate+ 0.4% Borax) | 0.93 | 50.95 | 49.04 |
| T6 (0.6% Calcium Nitrate + 0.8% Zinc Sulphate) | 0.84 | 60.30 | 39.70 |
| T7 (0.3% Calcium Nitrate + 0.2% Borax +0.4% Zinc Sulphate) | 0.94 | 40.29 | 59.71 |
| T8 (0.6% Calcium Nitrate + 0.4% Borax + 0.8% Zinc Sulphate) | 0.95 | 32.60 | 67.40 |
| CD (5%) | 0.07 | 2.82 | 2.63 |

Meena *et al.* 2014

Results

The increase in vegetative growth of plants might be due to stimulative effect of zinc because it is an essential element for chlorophyll formation which is directly related to photosynthetic activity of the plant (Khan *et al.*, 2009). Maximum fruit retention and minimum fruit drop might be due to calcium and boron being main constituent of cell wall (middle lamella) of plant cell in the form of calcium pectate which play an important role in strengthening of pedicel attached to proximal end of fruit resulted less fruit drop. Similarly reduction in fruit drop by spray of borax can be due

to the indirect action of boron in auxin synthesis that delayed the formation of abscission layer during early stages of fruit development (Guardiola and Garcia, 2000) [14]. According to Krishnamoorthy (1992) fruit drop is an abscission phenomenon controlled by the inter play of hormones. Zinc application at higher level increased the foliar zinc content which ultimately encourages the endogenous production of auxin thereby reducing fruit drop. Zinc is required for the synthesis of tryptophan a precursor of auxin thus helps in reducing fruit drop.

Table 7: Effect of micronutrients on fruit set and quality of litchi.

| Treatment | Fruit weight (g) | Fruit size (cm) | | L:D ratio | T.S.S. (%) | Titratable TSS : acidity acid ratio (%) | |
|----------------------|------------------|-----------------|----------|-----------|------------|---|-------|
| | | Length | Diameter | | | | |
| Boron 100 ppm | 15.96 | 3.44 | 2.94 | 1.17 | 17.88 | 0.44 | 40.78 |
| Boron 200 ppm | 17.26 | 3.48 | 3.01 | 1.16 | 19.00 | 0.41 | 47.07 |
| Zinc 100 ppm | 18.17 | 3.60 | 3.06 | 1.18 | 18.75 | 0.42 | 45.29 |
| Zinc 200 ppm | 16.55 | 3.32 | 2.96 | 1.12 | 18.25 | 0.39 | 46.83 |
| Control | 15.47 | 3.29 | 2.93 | 1.10 | 17.75 | 0.43 | 41.05 |
| CD _(0.05) | 0.95 | 0.13 | 0.07 | - | 0.89 | NS | - |

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Results

Higher fruit set by zinc application might be due to its effect on processes of fertilization and hormonal metabolism. Zinc is known to be essential for auxin synthesis (IAA) as it is an activator of enzyme tryptophan synthetase. Boron in higher-plants, has a crucial role in flowering, pollen germination and fruiting (Gauch and Dugger, 1954)^[12], it also increases the pollen producing capacity of the anthers and pollen viability, thus finally leads to higher fruit set. The possible reason for increase in fruit weight by the micro-nutrients and PBRs might be due to faster loading and mobilization of simple sugars into fruits and involvement in cell division and cell expansion (Brahmachari *et al.*, 1997). Increase in TSS content with these PBRs and micro-nutrients may be attributed to the quick metabolic transformation of starch and pectin into soluble compounds and rapid translocation of sugars from leaves to developing fruits (Brahmachari and Rani, 2001)^[5].

Conclusion

Micronutrients are required in very small quantities by the plant for their function. Since they are involved in various enzymatic activities, their deficiencies causes malfunctioning of the plant activities. To manage these micronutrient deficiencies spraying of suitable chemicals at recommended levels by foliar application will alleviate the deficiency. Increases in crop yields from application of micronutrients have been reported in many parts of the world. Factors such as pH, redox potential, biological activity, SOM, cation - exchange capacity, and clay contents are important in determining the availability of micronutrients in soils. Further, roots -induced changes in the rhizosphere affect the availability of micronutrients to plants. Major root induced changes in the rhizosphere are pH, reducing capacity, redox potentials, and root exudates that mobilize sparingly soluble mineral nutrients. Root exudates may make elements like Fe more available, but they may also produce water - soluble metal chelating agents which reduce metal activity with roots. Micronutrient application rates range from 0.2-100kg/ha, depending on the micronutrient, crop requirement and method of application. Higher rates are required for broadcast than for banded applications on soil or as foliar sprays. The development micronutrient-efficient and/or tolerant-resistant genotypes appear promising for improving future crop production.

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