



P-ISSN2349-8528
E-ISSN 2321-4902
IJCS 2016; 4(1): 14-17
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Received: 03-11-2015
Accepted: 04-12-2015

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Isolation and characterization of flavonoid glycoside from leaves of *Abrus precatorius*

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Abstract

In the present work, flavonoid glycoside compound was isolated from the leaves of *Abrus precatorius* and characterized by using thin layer chromatography. By using spectroscopic technique their structure and chemical bonds were analyzed. Phytochemical studies reveal the presence of apigenin flavonoid glucoside i.e., Vitexin.

Keywords: *Abrus precatorius*, flavonoids, TLC, NMR etc.

Introduction

The popularity of complementary medicine has increased in recent years. Dietary measures and traditional plant therapies as described by ayurvedic and other indigenous systems of medicines are used commonly in India. The natural compounds are not considered as chemicals and they are readily accepted by the consumers. Also, they have a GRASS (Generally Recognized as Safe) status and do not require toxicological testing. Flavonoids are major components of medicinal plants and have been used in traditional medicine around the world. *Abrus precatorius* Linn. Belongs to family fabaceae is an indigenous plant found all throughout India from Himalayan region to down the southern India. It is known as Gunja in Sanskrit, Gunchi in Hindi, Jequirity and Crab's eye in English. Commonly it is known as 'Ratti' or Rosary Pea'. Its seeds have remarkably uniform weight of 1/10th of a gram. So its seeds are used by goldsmiths; in old time to weight gold and silver. The plant has been used in Hindu medicines from very early times, as well as in China and other ancient cultures [1]. *Abrus precatorius* possess various pharmacological activities such as, antioxidative [2], antidiabetic [3], antiviral [4], antidepressant [5], memory enhancing [6], antimicrobial [7-12], antimalarial [13], anti-inflammatory [14-16], antiarthritic [17-18], anticancer [19], antifertility [20-26], antiallergic [27], antiasthmatic [28], anticataract [29], antiinsecticide [30], antitoxicity activity [31-32]. Leaves of *Abrus precatorius* resemble tamarind leaves having 20-40 leaflets. Its leaves are used in treating several diseases. The main object of this study is to extract and characterize flavonoid in the leaves of *Abrus precatorius*.

Experimental

¹H NMR and ¹³C NMR spectra were recorded on a Bruker Avance 400 MHz spectrometer. The EI-mass was recorded on Shimadzu QP 2000 mass spectrometer. UV-spectra were recorded on Shimadzu UV-160 spectrophotometer. The leaves of *Abrus precatorius* was washed thoroughly with tap water followed by rinsing with double distilled water and shade drying for fifteen days. The fine powder was obtained from dried leaves by using kitchen mixer grinder (Philips electronics). The leaves powder was sterilized at 120°C for 15 minutes. The leaves powder was stored under dessicator for further studies. Solvent extraction of dried powder (25gm) of *Abrus precatorius* was done using 250ml. of 80% methanol in a soxlet extractor for 36 hours. The extract was concentrated by evaporation (40°-50°) in vacuum rotatory. The concentrated methanolic extract (10ml.) was suspended in 50ml. of distilled water and was further extracted twice with hexane and then with ethyl acetate. The ethyl acetated fractions were washed two times with distilled water. The ethyl acetate fraction was analyzed for flavonid using chromatographic separation. The glass plates (20x20cm) coated with silica gel (0.2-0.3mm) were dried naturally (atmosphere). Subsequently they were activated at 100°C for 30 minutes and were cooled at temperature 25°C. Diluted samples of

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leaves of *Abrus precatorius* were qualitatively studied by TLC, butanol: acetic acid: water (4:1:5) upper layer was used as mobile phase. TLC plates coated with silica gel were used as stationary phase. The plates were sprayed with a solution of 1% ethanolic 2- amino ethyl diphenyl borinate followed by 5% ethanolic solution of polyethylene glycol-400. Flavonoid appears in color zone under UV-365nm. Standard flavonoids were used for identification. Retention time is 0.65. With both reagent A and B light green color is obtained. The remaining extract was evaporated and residue was obtained, it was subjected to various physical and spectral analyses.

Chemical Identification of flavonoids ^[33]

- 1. Shinoda Test:** To the small amount of extract in alcohol, magnesium ribbon was added followed by addition of drops of concentrated hydrochloric acid, formation of pink color confirms the presence of flavonoids.
- 2. Zn- Hydrochloride Reduction Test:** To the extract add a mixture of zinc dust and concentrated hydrochloric acid. Heat the solution, after few minutes, color of the solution changes to red.
- 3. Aluminium Chloride Test:** To the small amount of extract, two drops of 1% aluminum chloride was added, yellow color was obtained.

Results and Discussion

The compound was isolated as yellow amorphous powder m.p 203°C; ¹HNMR (DMSO-d₆), ¹³CNMR(DMSO-d₆) Table1. The ¹HNMR spectrum showed a doublet proton at δ 8.02 corresponding to H-2' and H-6' proton. Another doublet proton occurs at δ 6.88 corresponding to H-3' and H-5'. Two

protons appeared at δ 6.60 and δ 6.26 as singlets corresponding to H-3 and H-6 protons respectively. The ¹³CNMR spectrum of the compound showed 21 signals for the vitexin. Carbon bonded to the carbonyl group C-4 appeared at δ 182.0. The carbonyl carbon, C-4 resonates around δ 175-178, when the carbonyl is not hydrogen bonded. But in the presence of H-bonding to 5-hydroxy group, it moves downfield to about δ 182. When 3-hydroxy group is alone it resonates at δ 171-173. When both 3- hydroxyl and 5-hydroxyl groups are present, it resonates at δ176. Carbon bonded to the hydroxyl group C-5, C-7 and C-4' appeared at δ 61.1, δ 162.4, δ 160.2 respectively. Signals of C-6 from C-8 and signals of C-5 from C-9 are distinguished with the help of ¹³C-¹H coupling data. The degree of coupling identifies each carbon and demonstrates that C-9 resonates at higher field from C-6 while C-8 resonates at higher field from C-6. The degree of coupling identifies each carbon and demonstrates that C-9 resonates upfield from C-5 while C-8 resonates up field in comparison to C-6.

The UV spectrum of this compound exhibited two major peaks in the region 335nm and 271nm which indicates the presence of flavonoid structure. Mass spectra of isolated compound show molecular ion m/z 433 [M⁺] corresponding to the molecular formula C₂₁ H₂₀ O₁₀. The presence of vitexin in *Abrus precatorius* was not reported before the present study. The vitexin was reported earlier from leaves of *Parkinsonia aculeate* ^[34], *Ficus deltoidea* ^[35], *Alysicarpus monilifer* ^[36], *Crataegus pinnatifida* ^[37], etc. From the above studies it was concluded that this compound is 5,7,4'-trihydroxy flavones -8-glucoside i.e., apigenin 8-C glucoside or vitexin.

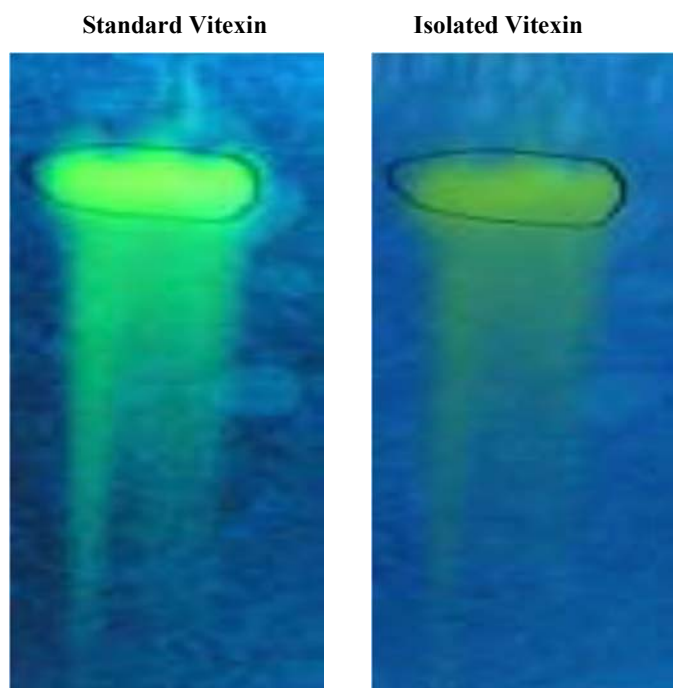


Fig 1: Flavonoid finger printing of *Abrus precatorius*

Vitexin possess hepatoprotective, antidiabetic ^[36], analgesic ^[40] activity. Vitexin is able to protect nerve cells against oxidative injury ^[38]. Vitexin is antineoplastic agent, platelet aggregation inhibitor possess neuroprotective properties ^[39]. It has a long term effect and improved the neurobehavioral outcomes. Vitexin is HIF-1α inhibitor. It is natural flavonoid antioxidant and may be useful in treatment of various diseases. The

suspicious carcinogenic effect of synthetic antioxidants such as BHA and BHT has raised significant concerns about the use of chemical additives in food. The possible mechanism of action of bioactive flavonoid glycosides vitexin could be to block the active sites of reactive oxygen species to which they have greater affinity and cause cellular damage, there by offering protection to the targeted tissue acting as antioxidant.

Flavonoids have been identified as a new type of neuromimetic ligand with *in vivo* possess anxiolytic properties [41]. Flavonoid glycosides, showed to exert central nervous system mediated activities, particularly as sedative-hypnotics, analgesics and anxiolytic [42-43].

Thus, recently the demand of natural antioxidants has increased enormously and production of vitexin from leaves of *Abrus precatorius* may be useful if it is used in place of synthetic antioxidants. This study confirms flavonoid vitexin in the leaves of *Abrus precatorius*.

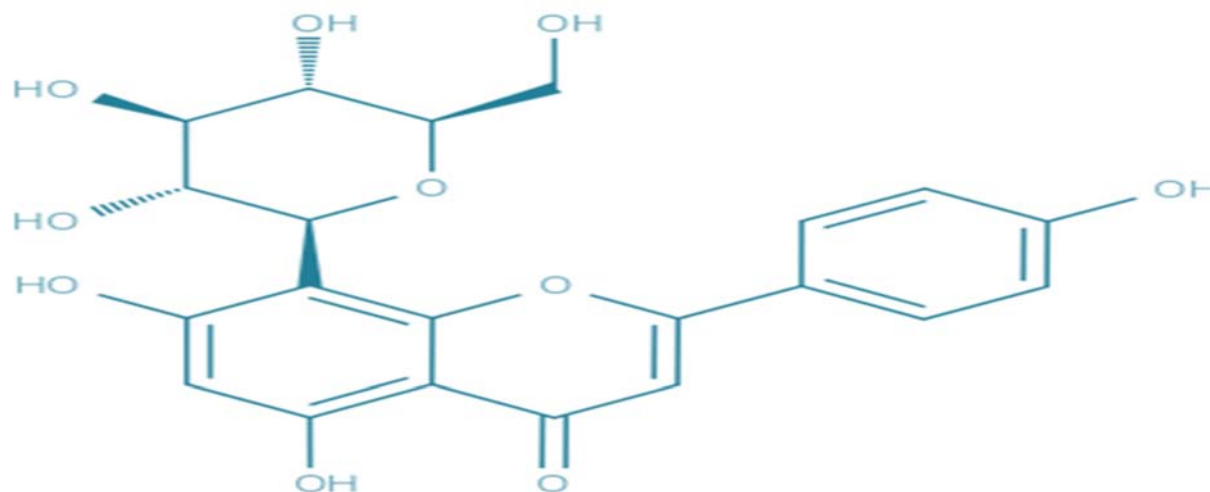


Fig 2: Vitexin

Table 1

¹ H NMR Spectral Data for Compound (400MHz, DMSO, δ(ppm))	¹³ C NMR Spectral Data for Compound (100 MHz, DMSO, δ (ppm))
H- Position	C-position
3 (6.60)s	163.8(C-2)
6 (6.26)s,	102.3(C-3)
2' (8.02) d J=8.4	182.0(C-4)
3'(6.88)d J= 8.4	161.1(C-5)
5'(6.88)d J= 8.4	98.2 (C-6)
6' (8.02)d J=8.4	162.4(C-7)
1'' (4.68)d	104.5(C-8)
	155.8(C-9)
	104.0 (C-10)
	121.4(C-1')
	128.7(C-2')
	115.6(C-3')
	160.2(C-4')
	115.6 (C-5')
	128.7 (C-6')
	73.2(C-1'')
	70.8 (C-2'')
	78.6(C-3'')
	70.4(C-4'')
	81.5(C-5'')
	61.8(C-6'')

Acknowledgements

I am very grateful to University Grants Commission, New Delhi, India for their financial assistance (Grant No.F.15-39/12 (SA-II)). I am also very thankful to Dr. A.K. Gupta Head of the Chemistry Deptt. and Dr. M.K. Rawat, Principal Agra College, Agra, for their full support.

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